- 1. What is the smallest unit of matter? (atom)
- 2. What are the three subatomic particles that make up an atom? (proton, neutron, electron)
- 3. Which subatomic particle has a negative charge? (electron)
- 4. What is the difference between a proton and neutron? (proton + neutron 0 neutral)
- 5. Where are electrons located within an atom? (Orbiting the nucleus)
- 6. What subatomic particles can be found in the nucleus? (protons & neutrons)
- 7. What is the charge of the nucleus? (positive)
- 8. If an atom gains an electron, what is its charge? (negative)
- 9. What subatomic particle is positively charged? (proton)
- 10. What kind of charge does a neutron have? (trick question: neutrons are neutral)
- 11. What kind of bond is formed when electrons are transferred from one atom to another? (ionic)
- 12. What is formed when an atom gains or loses an electron? (ion)
- 13. Name one of the four primary elements essential for life. (carbon, hydrogen, nitrogen, oxygen)
- 14. When 2 hydrogen atoms bind to an oxygen atom, what molecule is formed? (water)
- 15. In a water molecule, what kind of bond is formed between the hydrogen and oxygen atoms? (covalent)
- 16. Water is a polar molecule meaning there is an uneven distribution of electrons. Which part of a water molecule has a slight negative charge? (oxygen)
- 17. When two atoms share electrons, what kind of bond is formed? (covalent)
- 18. If an atom of Na loses an electron and a Cl atom gains an electron, what type of bond will be formed between the two atoms? (ionic)
- 19. When one water molecule binds to another water molecule, what kind of bond is formed? (hydrogen)
- 20. Water molecules are "sticky". What property allows them to bind easily to one another? (cohesion)
- 21. Water is able to travel through the roots and up to the very top of a tree because of what two properties? (cohesion and adhesion)
- 22. The tendency of unlike molecules to stick to one another is called what? (adhesion)
- 23. In a polar molecule, which subatomic particle is not shared equally? (electron)
- 24. What does -phobic mean/refer to? (afraid/repel)
- 25. What does -philic mean/refer to? (loving)
- 26. What does hydro- mean/refer to? (water)
- 27. What about oil, and other like substances, allows it to sit on top of water? (hydrophobic: ionic or polar substance that repels water)